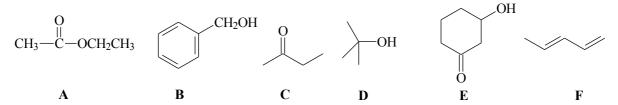
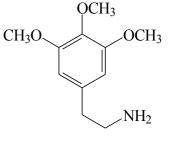
CHEM1002 Problem Sheet 4 (Week 4)

Review 'E35 Spectroscopic Techniques in Organic Chemistry' in the Week 3 Resources.

1. Answer the following questions concerning the six compounds below.

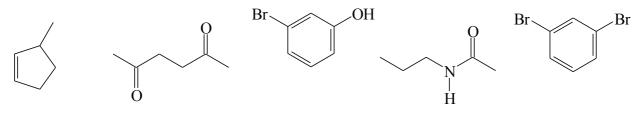


- (a) Which compound(s) will give a molecular ion at m/z = 74 in the mass spectrum?
- (b) Which compound(s) will show a strong absorption in the UV-visible spectrum?
- (c) Which compound(s) will show absorption around 1700 cm⁻¹ in the infrared region?
- (d) Which compound(s) will show absorption around 3500 cm^{-1} in the infrared region?
- (e) Which compound(s) will *not* show absorption either around 1700 or 3500 cm^{-1} in the infrared region?
- (f) How many signals will each compound have in its ¹³C NMR spectrum?
- 2. A compound of formula C_7H_6O has a molecular ion at m/z = 106 in the mass spectrum, shows a strong absorption in the UV region and an absorption around 1700 cm⁻¹ in the IR. Suggest a structure for the compound.
- 3. Mescaline is a hallucinogenic compound produced by the peyote cactus.
 - (a) Name the functional groups present.
 - (b) How many signals will be observed in its ${}^{13}C$ NMR spectrum?
 - (c) At what value of m/z will it show a molecular ion (parent ion) in the mass spectrum?
 - (d) Do you expect it to have a strong absorption in the UV-visible spectrum?
 - (e) Do you expect it to have a strong absorption around either \sim 3400 or \sim 1700 cm⁻¹ in the infrared spectrum? If so, what functional groups is/are responsible for this absorption?

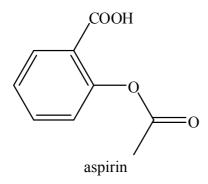


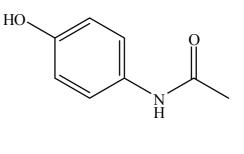
Mescaline

4. How many signals will be observed in the ¹³C NMR spectrum of each of the following compounds?



5. By considering both the number and type of carbon environments, *sketch* the ¹³C NMR spectra you would expect for aspirin and paracetamol. (You do *not* need to worry about the relative position of the lines due to the aromatic ¹³C nuclei).





paracetamol

6. Name the functional group(s) in each of the following compounds, including classification into primary, secondary or tertiary where appropriate.

